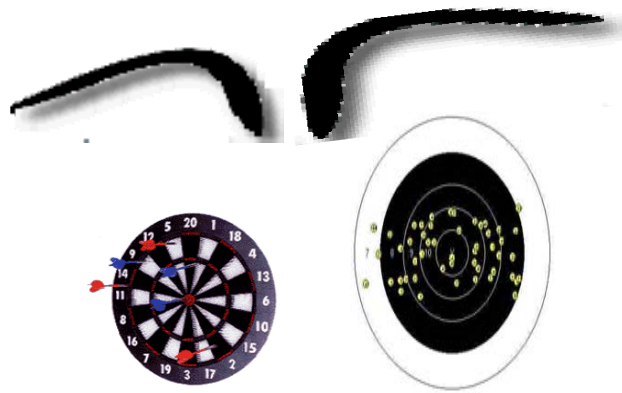




# Targets for nuclear physics studies



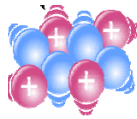
*Anna Stolarz*

**Heavy Ion  
Laboratory**

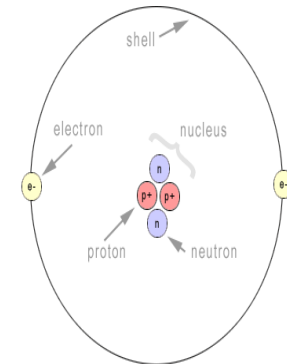
*What is the target?*



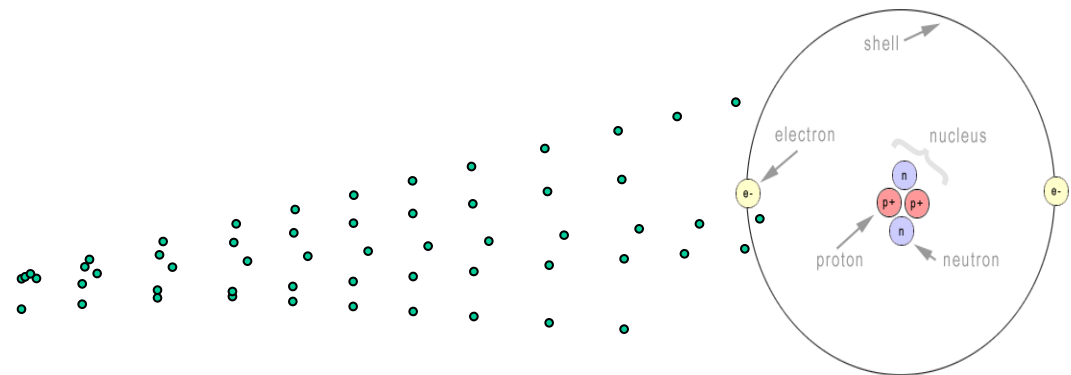
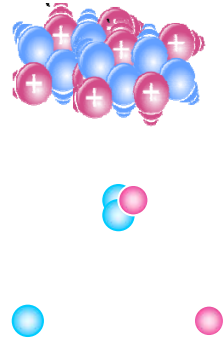
Nuclear reaction is a process in which two, three nuclei or nuclei and particle such as neutron, proton ... collide producing the product other than the initial 'items'.



+



Nuclear reaction is a process in which two, three nuclei or nuclei and particle such as neutron, proton ... collide producing the product other than the initial 'items'.



Lithium 7  
(Li-7)



**The first artificial nuclear reaction was performed by John Cockcroft and Ernest Walton in 1932. They bombarded the  ${}^7\text{Li}$  with ‘artificially’ accelerated protons. In result the two helium nuclei ( $\alpha$  particles) were created.**

## *What is the target?*

### **Gasous or liquid**

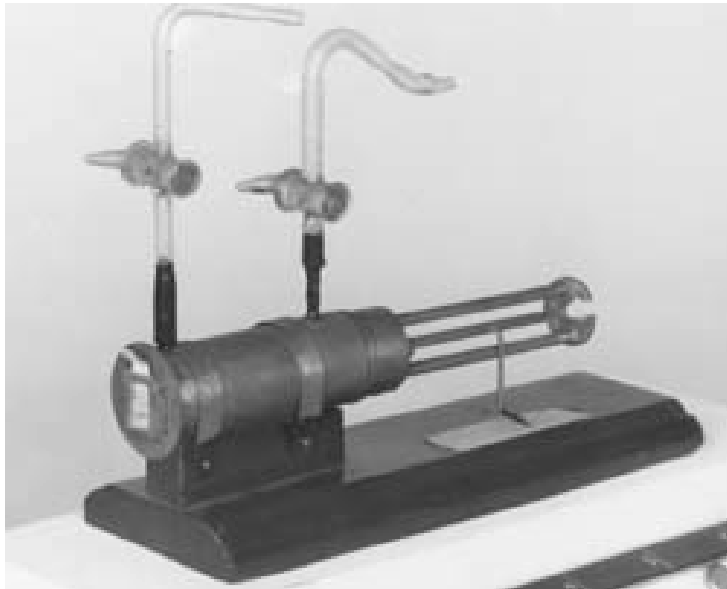
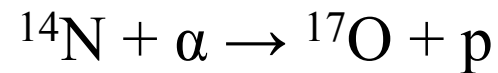
- gas or liquid flow (the melted metals as well)
- material closed in the chamber kept in the low temperature
- in case of gaseous target:  
implantation into the solid backing/carrier



*What is the target?*

## Rutherford transmutation „An anomalous effect in nitrogen,”

the alpha particle (from Polonium), which passed through the container with nitrogen gas, and nitrogen nucleus stuck together with a proton flying loose.

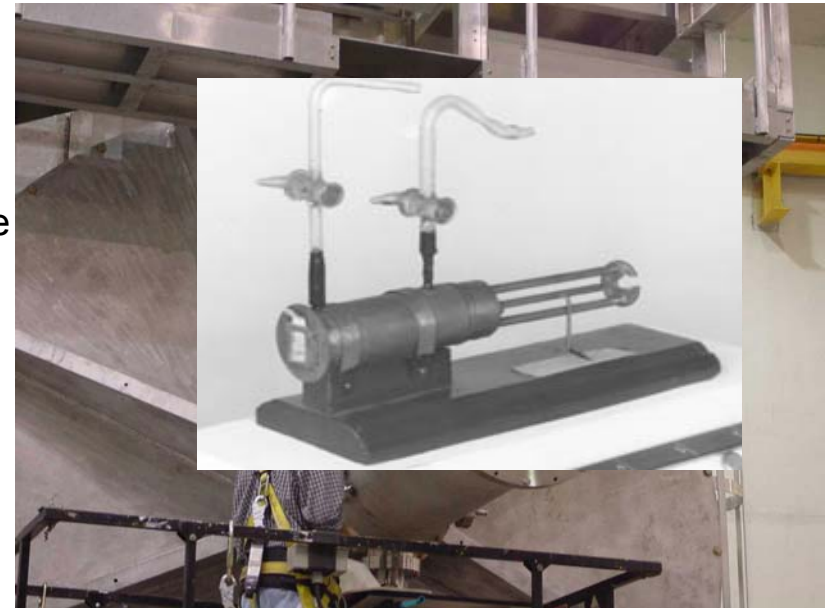


The apparatus used in 1919 by Rutherford's team for observation of the  $\alpha$  particles interaction with light nuclei what resulted with transmutation of nitrogen into oxygen

## What is the target?

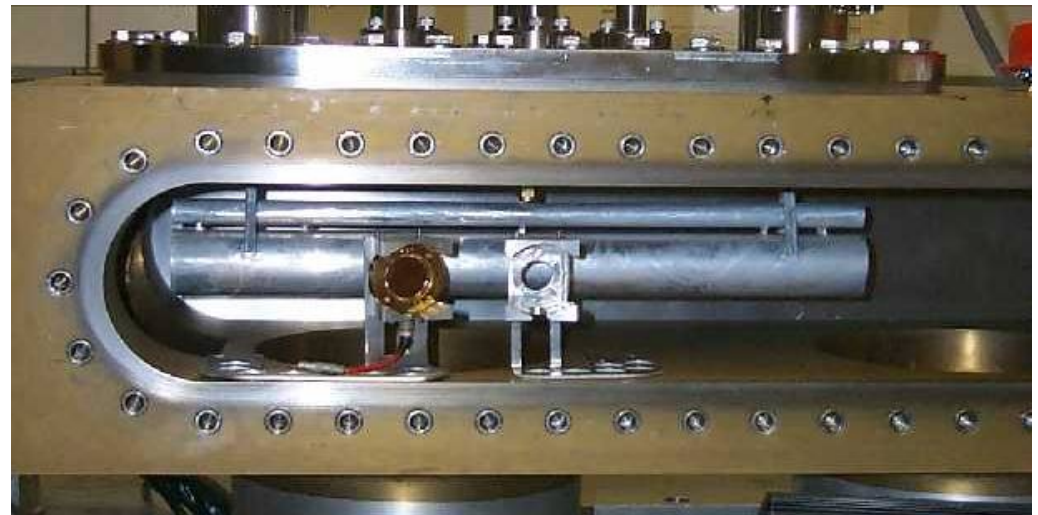
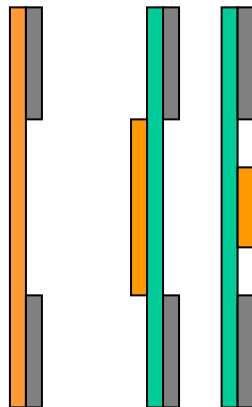
### Gasous or liquid

- gas or liquid flow (the melted metals as well)
- material closed in the chamber kept in the low temperature
- in case of gaseous target:  
implantation into the solid backing/carrier



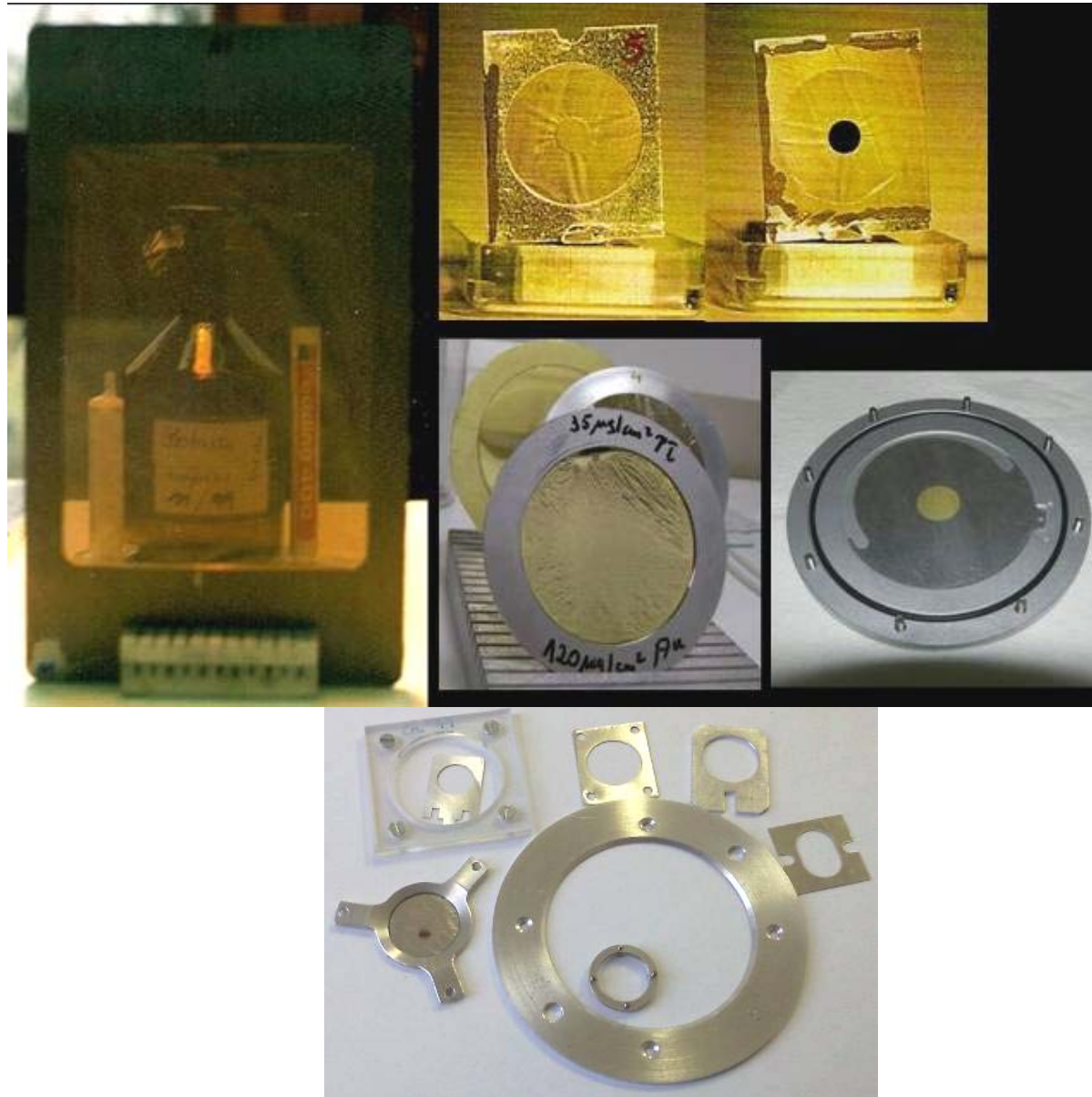
### Solid targets

- self supporting
- on the backing





*What is the target?*



Workshop, February 2013

## *How to make the target?*

The choice of the method depends on many aspects:

- target form (phase) and characteristics/parameters:  
element/isotope, thickness, size
- availability of the tools/method in the target lab
- avoiding unnecessary costs
- avoiding contamination of the material

# *Target properties/characteristic*

Target material: element-isotope  
and its phase: solid, liquid, gaseous

Thickness and its homogeneity

Chemical form required and available

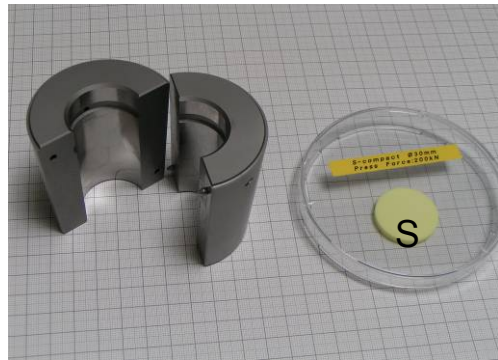
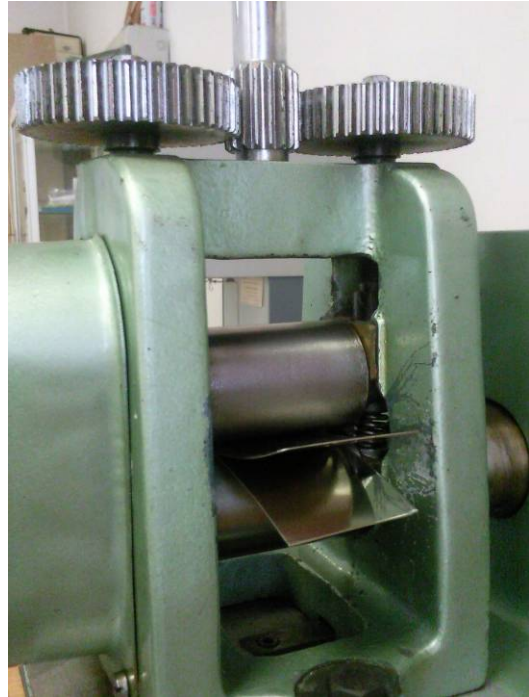
Self-supporting or on the backing

*How ???*

mechanical shaping:

rolling

tablet pressing



*How ???*

chemically: electro-deposition from hydrous or organic medium  
(always on the backing)



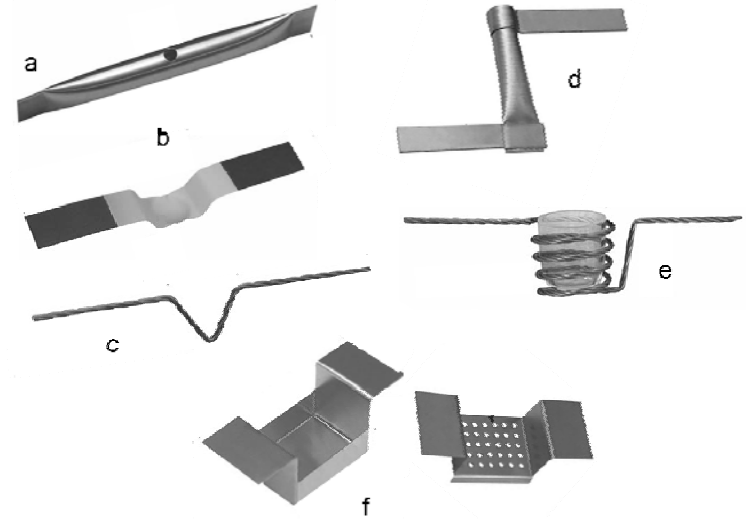
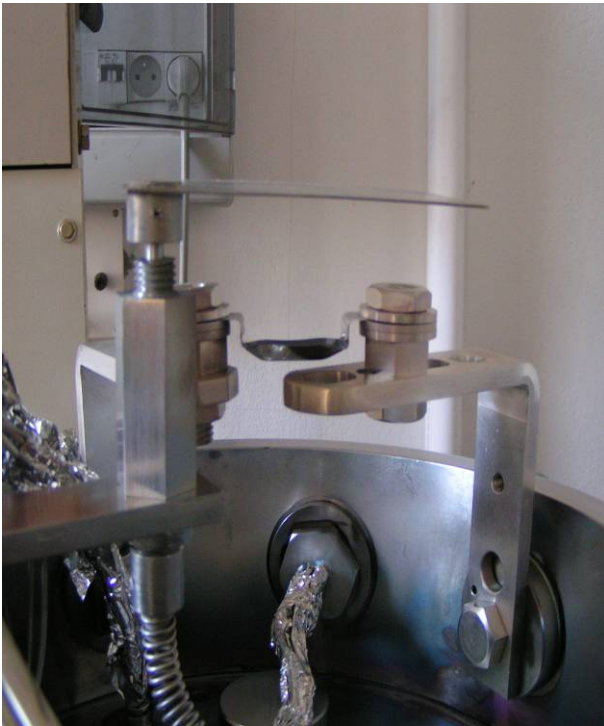
*How ???*

vapour deposition in the high vacuum  
(self-supporting or on the backing)

-resistance heating

-e-gun

-sputtering



*How ???*

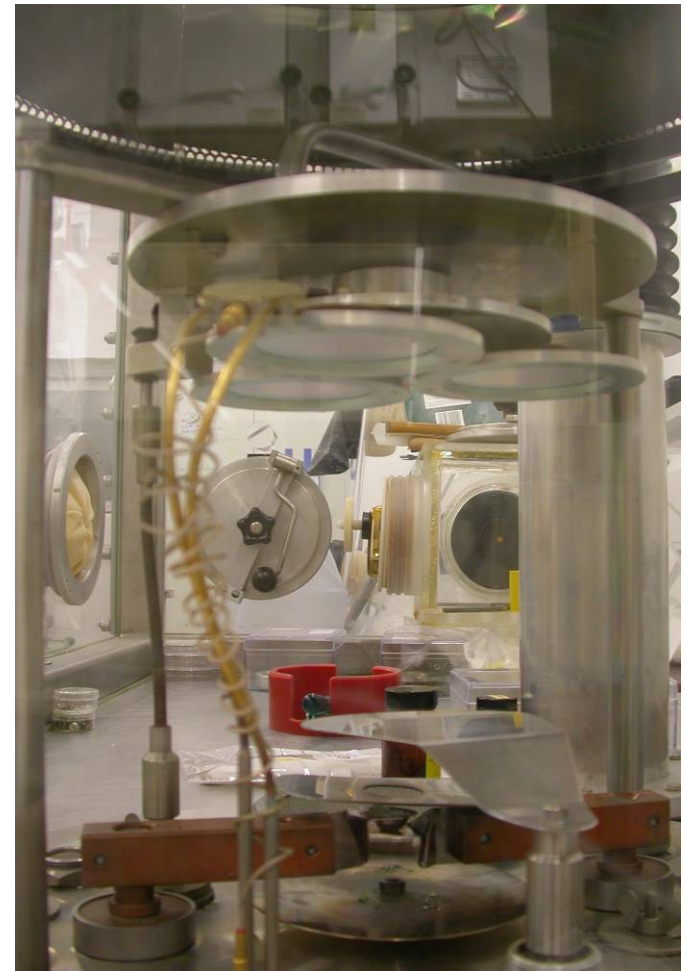
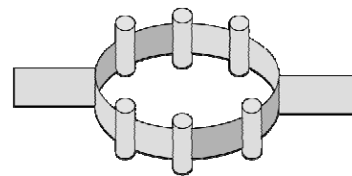
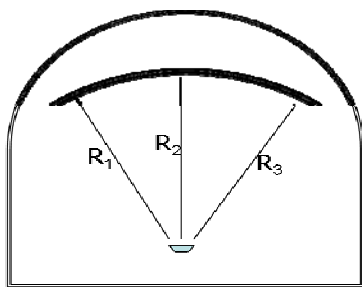
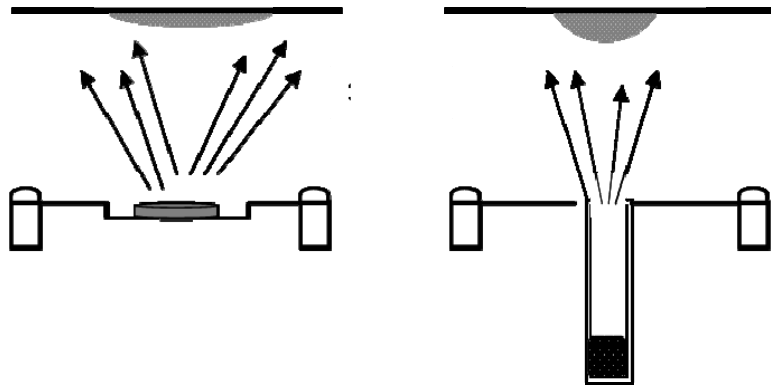


sputtering i.e. target material  
ejection by accelerated ions of the  
nobel gas



*How ???*

vapour deposition in the high vacuum





*How??*

## Carbon foil

carbon arc  
laser ablation  
e-gun  
sputtering



*How ???*

## Resistance heating

- The method is very simple, robust

**but**

- limited to the materials of the low melting point (not higher than 1800 °C)
- and not alloying with the boat material.

## E-gun

- The method is more complex, but extremely versatile.
- Can achieve temperatures in excess of 3000°C.
- Use evaporation cones or crucibles in a water cooled copper hearth.
- Typical emission voltage is 8-10 kV.

**but**

- Exposes substrates to secondary electron radiation.
- X-rays can also be generated by high voltage electron beam

## Sputtering

- The method can be applied to the most of the materials except those which can degrade due to ionic bombardment
- This technology allows to release the deposited material at much lower temperature than evaporation.
- gives easy film thickness control via time, allows alloy deposition, no x-ray damage

**but**

- requires rather big surface of the sputtered material to avoid bombarding of the cathode material.
- There is as well big chance for the impurities incorporation due to low vacuum.

# Backings?

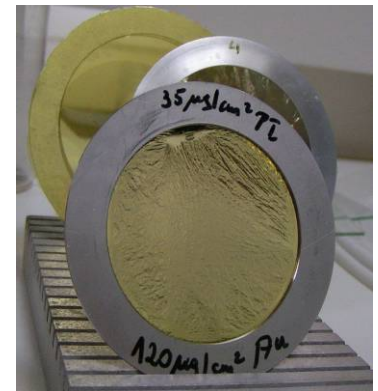
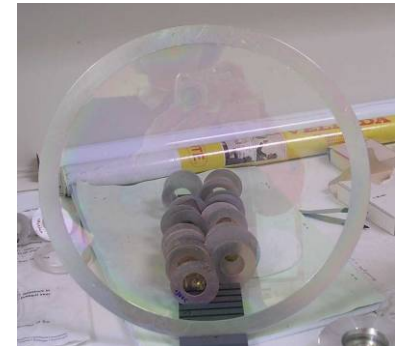
on the backing or self-supporting

backings

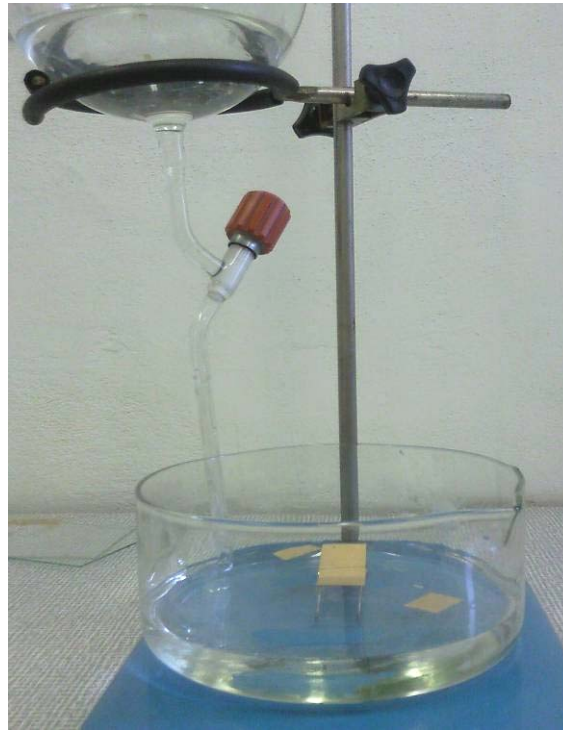
thin metal foils

carbon foil

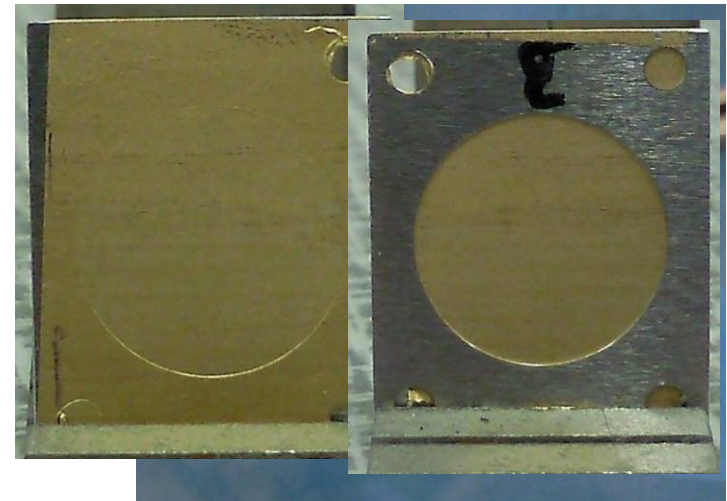
plastic: Mylar, Kapton, Formvar



How??  
?



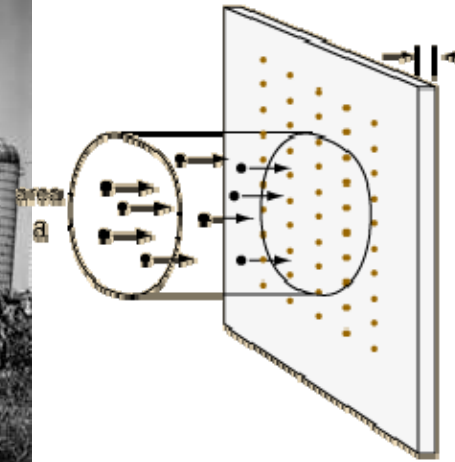
by vapour deposition  
on substrate



# Target characterisation

**Thickness:**

(mass/area i.e. g-mg- $\mu$ g/cm<sup>2</sup>)

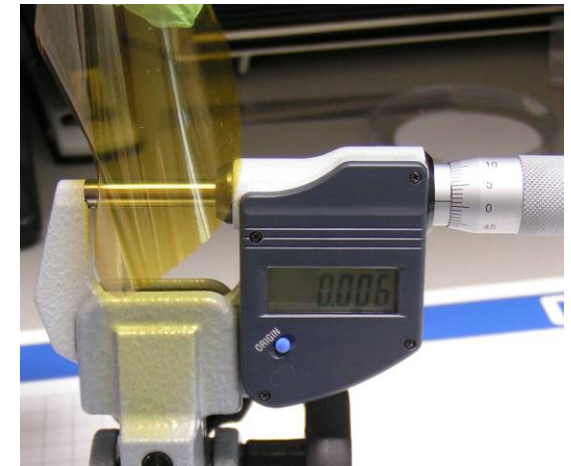


$$1 \text{ b } (\sigma) = 10^{-24} \text{ cm}^2$$

it's approx. the sectional area of the U nucleus

**Thickness estimation:** mass/area i.e. g-mg- $\mu$ g/cm<sup>2</sup>)

- \* mechanically or electrically i.e. using caliper, micrometer screw or thickness induction gauge
- \* weighing the defined area
- \* in-situ during the vapour deposition process using the quartz microbalance
- \* spectrophotometrically
- \* measurement of the  $\alpha$  particles or X-ray energy
- \* profilometers  
working in a contact or non-contact modes

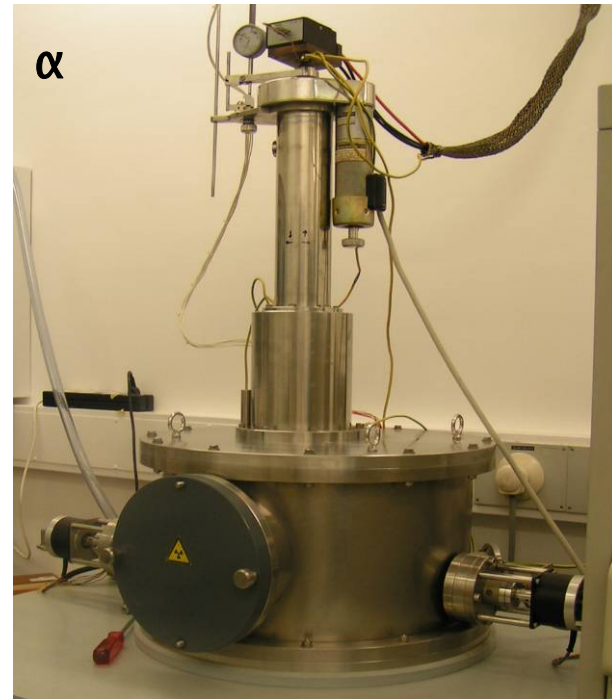


## *Thickness estimation of the radioactive targets:*

*if made by evaporation:* during preparation with quartz microbalance

*ready target:* measurements of the radioactivity

thickness homogeneity by radioactivity scan across the target area

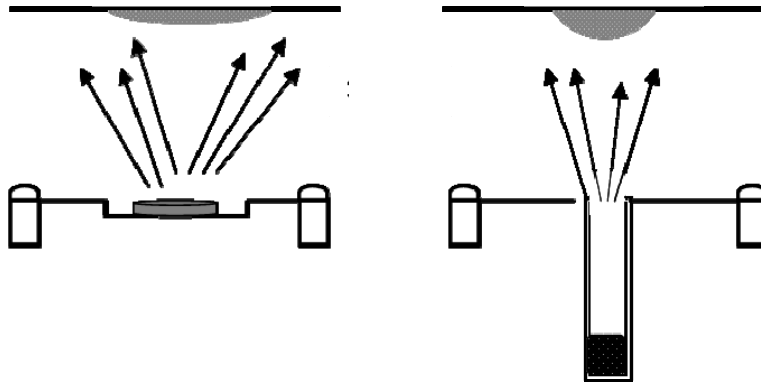


# Target characterisation

**Thickness:** (mass/area i.e. g-mg- $\mu\text{g}/\text{cm}^2$ )

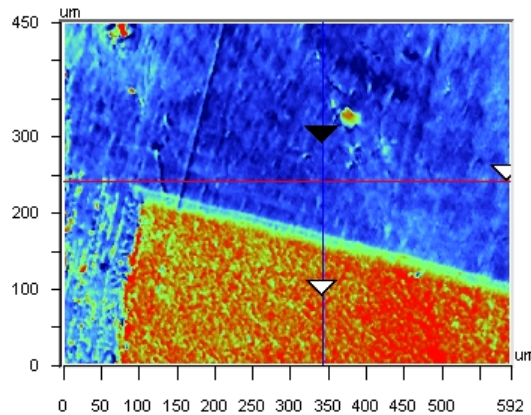
**Thickness homogeneity:**

?



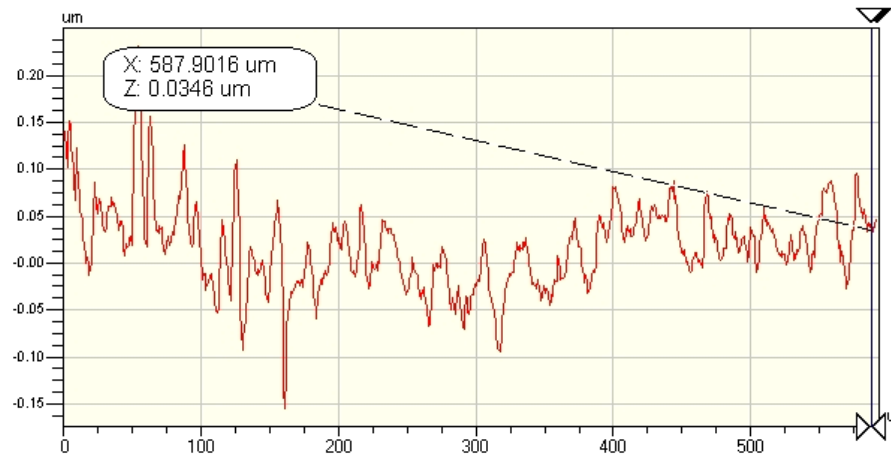


# Surface characterisation



X	342.28	-	-	um
Y	240.53	-	-	um
Ht	-0.02	-	-	um
Dist		-	-	um
Angle		-	-	°

X Profile



Rq	0.05 um
Ra	0.03 um
Rt	0.39 um
Rp	0.23 um
Rv	-0.16 um

Angle	0.00 mrad
Curve	0.56 m
Terms	None
Avg Ht	0.02 um
Area	10.22 um <sup>2</sup>

Y Profile

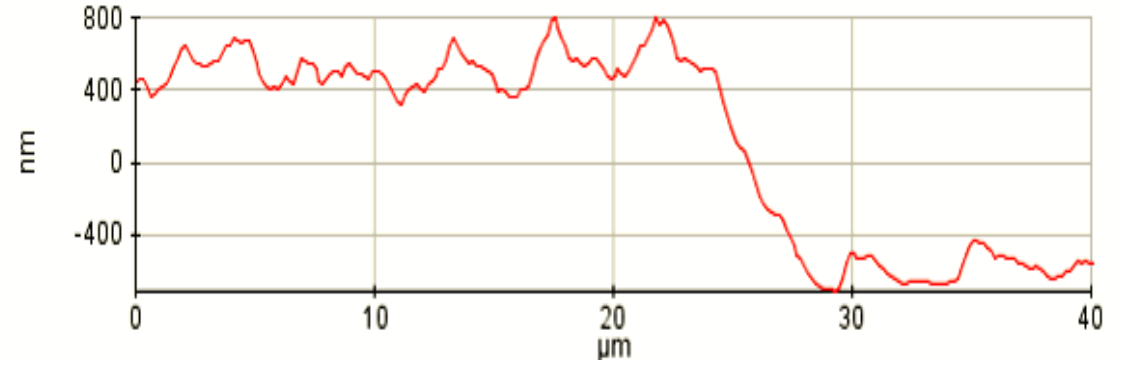
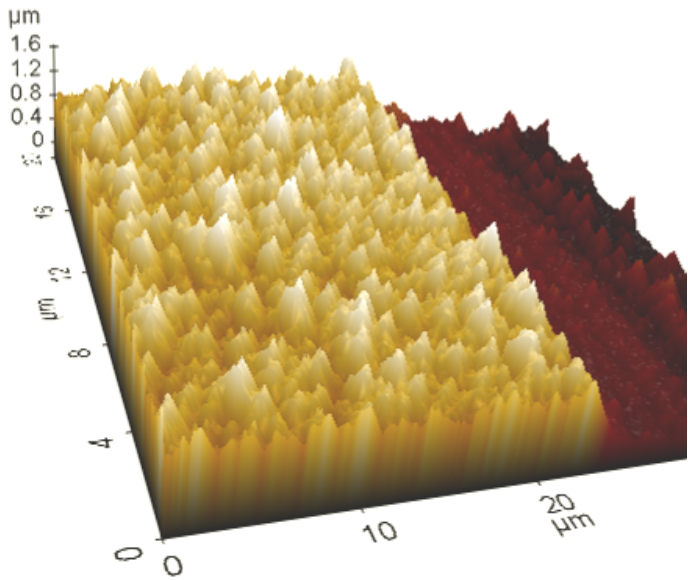


Rq	0.21 um
Ra	0.19 um
Rt	0.65 um
Rp	0.58 um
Rv	-0.07 um

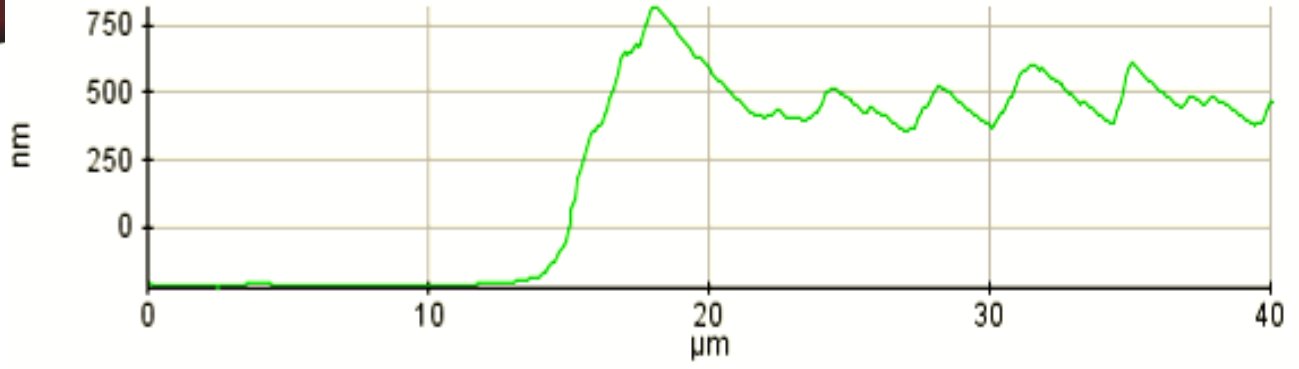
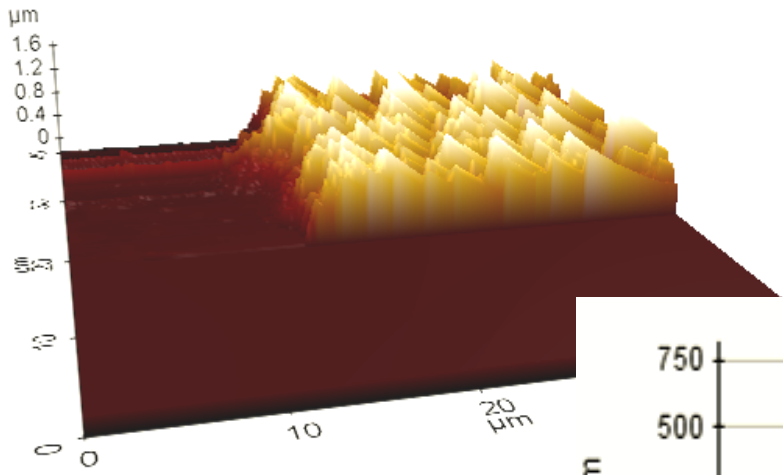
Angle	-2.31 mrad
Curve	22.77 mm
Terms	None
Avg Ht	0.16 um
Area	32.59 um <sup>2</sup>

Title:

## Surface characterisation



## *AFM images of tristearin layer*



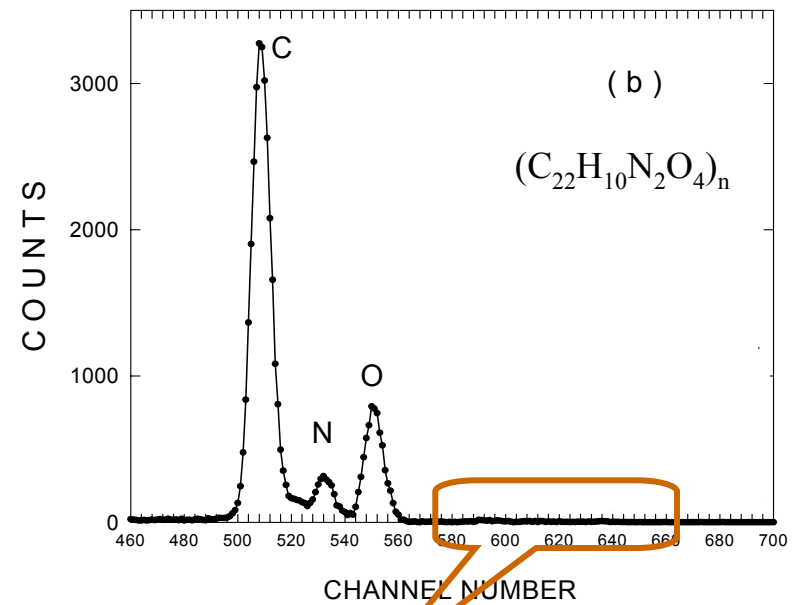
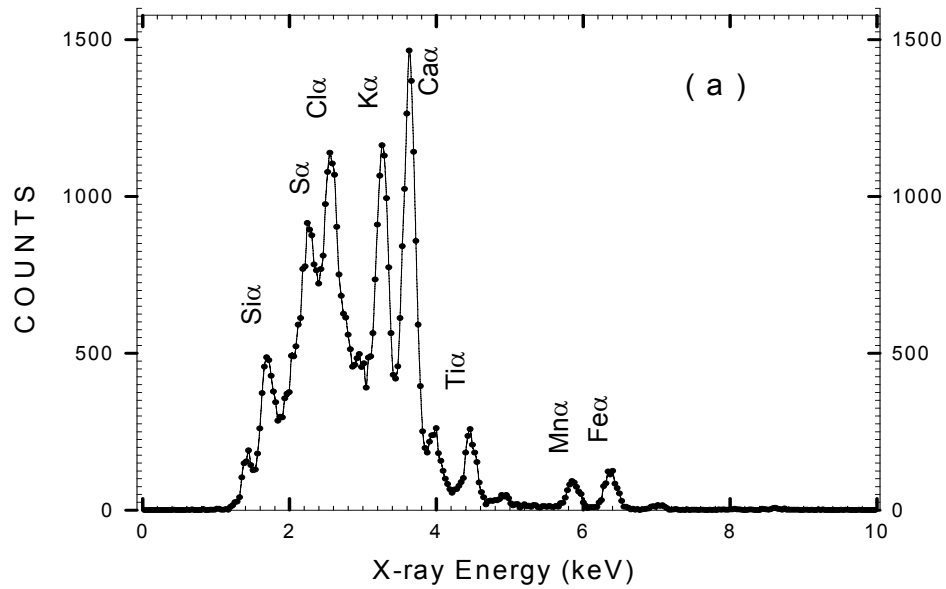
# *Target characterisation*

**Thickness:** (mass/area i.e. g-mg- $\mu\text{g}/\text{cm}^2$ )

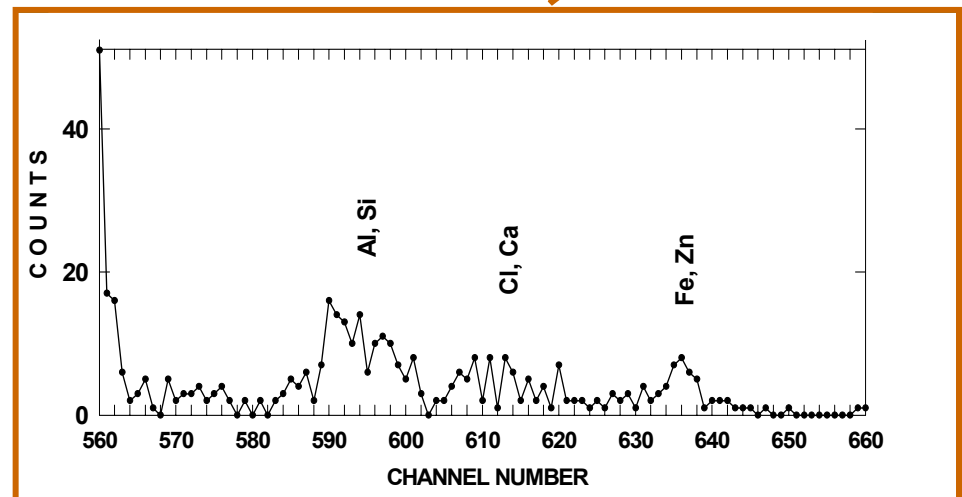
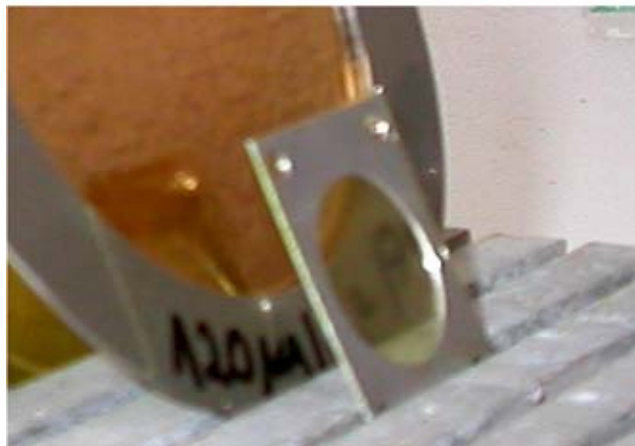
**Thickness homogeneity (including surface topography)**

**Purity/composition**

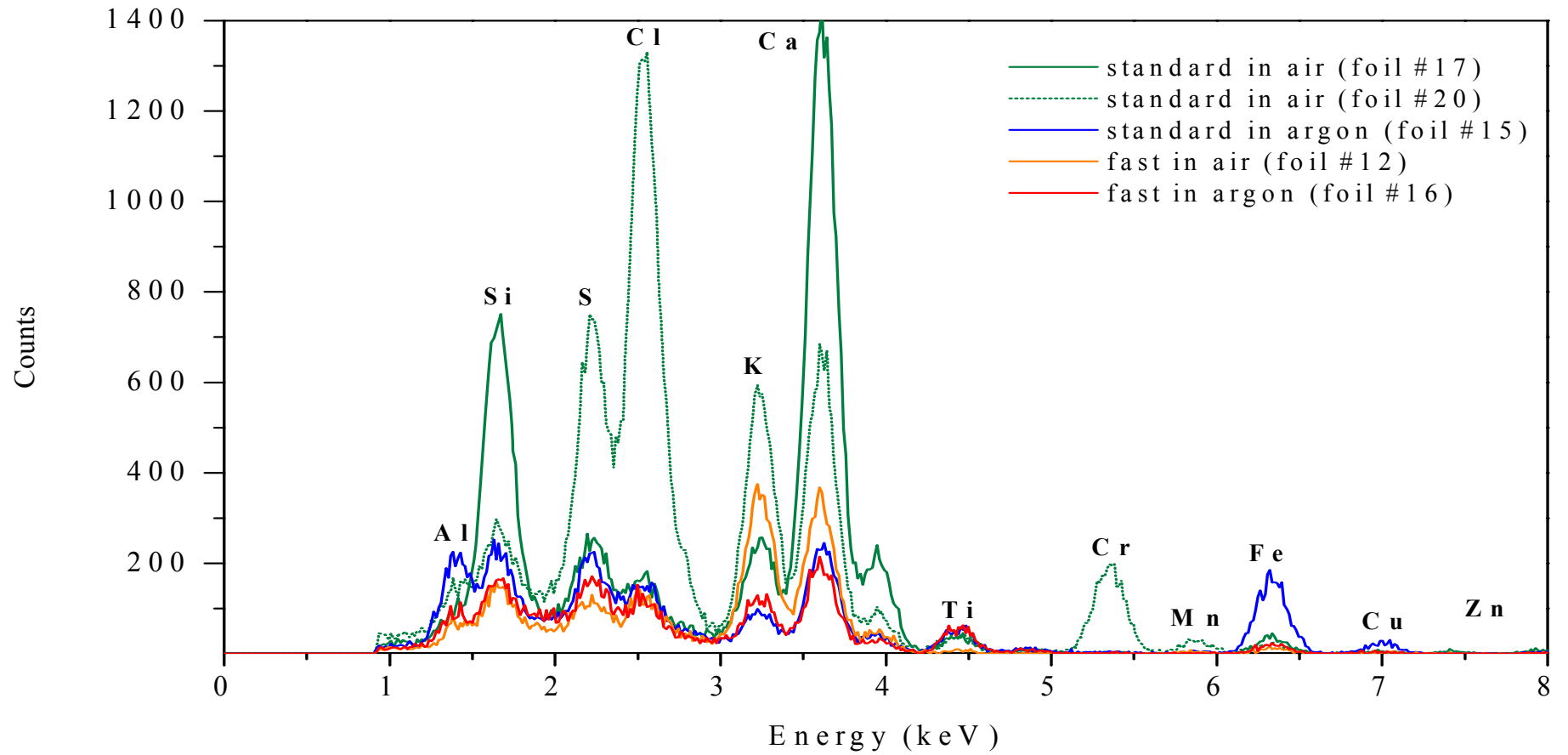
# Purity/composition



PIXE spectrum of polyimide foil



# Purity/composition



## *Closing remarks*

When ordering a target define the characteristic needed/significant for planned studies but avoid exaggeration i.e. do not order a target with much better characteristic than really needed. This may cause additional costs and/or ... delay.

*element/isotope*

*thickness, dimensions*

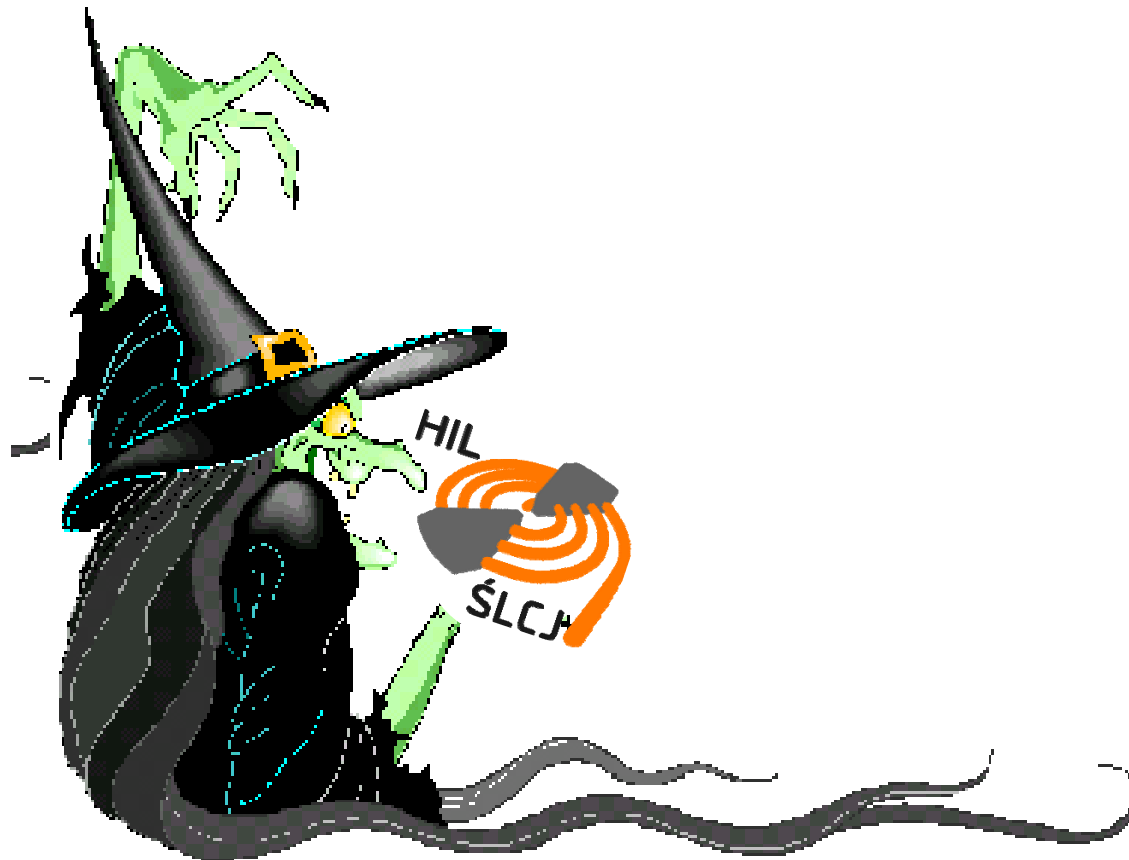
*supported or not, if yes what can be considered as support*

*purity*

Do not overestimate the importance of the chemical form of the target material.

*not always have to be a pure elemental form, the compounds may suite your needs as well but often it is much easier (cheaper) to make the target from compound*

Discuss with target maker your planned target. Target preparation people can do sometimes more for you than you believe; it is often a question of communication and of raising the relevant problems/aspects.



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[www.intds.org](http://www.intds.org)